

TETRAHEDRON

Ring-Closing Olefin Metathesis for the Synthesis of Phosphorus Containing Heterocycles

L. Hetherington, B. Greedy and V. Gouverneur*

University of Oxford, The Dyson Perrins Laboratory, South Parks Road, OX1 3QY Oxford, UK Received 24 September 1999; revised 6 January 2000; accepted 20 January 2000

Abstract—A series of cyclic phosphorus containing heterocycles 9a-k was prepared in a one-step procedure by ring closing metathesis of dienes 7a-k. © 2000 Elsevier Science Ltd. All rights reserved.

Introduction

Heterocyclic systems that include a phosphorus linked to an oxygen or nitrogen atom are common to a diverse array of important biological molecules, e.g. **1–5** (Scheme 1).

These include cyclophosphamide **1a**, an anti-tumor alkylating agent which together with its structural isomer ifosfamide **1b** act as prodrugs relying on metabolism for their action.¹ In another group of compounds, cyclic phosphonates as exemplified by compound **2** are hexapyranose analogs modified at the anomeric carbon.² As a result, these analogs can regulate key steps in carbohydrate linked biological processes, e.g. cellular recognition.³ 1,4-Dihydropyridine-5-cyclic phosphonate derivatives such as compound **3** are analogs of 1,4-dihydropyridine-3,5-dicarboxylate calcium antagonists and are therefore anti-hypertensive agents.⁴ 1,2-Azaphosphorine such as compound **4** were developed in the search for biodegradable insecticides.⁵ More recently, cyclic organophosphorus compounds have been employed for the generation of novel biocatalysts.⁶ The cyclic phosphonate **5** has been successfully used to generate antibodies that catalyse the enantioselective aminolysis of lactones.⁷

We were recently attracted by the versatility and synthetic applicability of the ring-closing metathesis reaction (RCM)⁸ using the well-defined transition metal catalysts $6a,b^9$ and $6c^{10}$ in the construction of functionalised carbocycles and heterocycles (Scheme 2).



Scheme 1.

Keywords: ring closing metathesis; phosphorus heterocycles.

^{*} Corresponding author. Fax: +44-1865-275-644; e-mail: veronique.gouverneur@chem.ox.ac.uk



Scheme 2.

To date, the literature contains only a few examples of RCM reactions on phosphorus containing compounds. These include a phosphine,¹¹ phosphinate,¹² phosphinate¹³ or phosphine oxide¹⁴ functionality. However, there exist no

comprehensive study that explores the scope and limitation of the RCM reaction to generate such phosphorus containing compounds. As part of a study to prepare new transition state analogs for antibody catalysis, we extended this methodology to the preparation of new phosphorus/oxygen heterocycles with a focus on 3,6-dihydro-1,2-oxaphosphinine 2-oxide and to the synthesis of phosphorus/nitrogen heterocycles.

Results and Discussion

To establish the feasibility of this strategy, a series of representative dienes were synthesised using three different procedures (Scheme 3).





condition: 2 eq allylic alcohol, 2 eq. Et_3N , cat DMAP**7a** X = Oyield: 98%condition: 5 eq allylamine, 2 eq. Et_3N , cat DMAP**7b** X = NHyield: 41%

Preparation of dienes 7c, 7d, 7e and 7f



8c $R^1 = R^2 = R^3 = H$	7c $R^1 = R^2 = R^3 = H$	overall yield: 31%
8d $R^1 = R^3 = H, R^2 = Me$	7d $R^2 = Me, R^1 = R^3 = H$	overall yield: 46%
8e $R^1 = R^2 = H, R^3 = Me$	7e $R^3 = Me, R^1 = R^2 = H$	overall yield: 45%
8f $R^1 = Me, R^2 = R^3 = H$	7f $R^1 = Me, R^2 = R^3 = H$	overall yield: 31%

Preparation of dienes 7g, 7h and 7i

$$\begin{array}{c} \bigcirc \mathsf{Ph} \\ \bigcirc \mathsf{OMe} \end{array} \xrightarrow{(1)1 \text{ eq. PCl}_5, \text{ rt}} \\ \hline (2) 3 \text{ eq } \mathsf{HXCHR}_1\mathsf{CR}_2 = \mathsf{CH}_2, 3 \text{ eq } \mathsf{Et}_3\mathsf{N}, \\ & \mathsf{cat } \mathsf{DMAP}, 0^{\circ}\mathsf{C} \text{ to } \mathsf{rt}, 72 \text{ h} \end{array} \xrightarrow{\mathsf{O}} \mathsf{Ph} \overset{\mathsf{Ph}}{\underset{\mathsf{R}^2}} \xrightarrow{\mathsf{O}} \mathsf{Ph}^{\mathsf{Ph}} \xrightarrow{\mathsf{Ph}} \overset{\mathsf{O}}{\underset{\mathsf{R}^2}} \\ \end{array}$$

 $7g X = O, R^1 = H, R^2 = Me$ overall yield: 39% $7h X = NH, R^1 = R^2 = H$ overall yield: 38% $7i X = NH, R^1 = Ph, R^2 = H$ overall yield: 33%

Preparation of dienes 7j and 7k

MeOOC

Entry	Substrate	Product	Condition	Yield ^a
1	Ph PR O 7c	Ph O Pc	4% 6a , 16 h	92%
2	w Ph PH PH PH Ph Ph 7d	Ph O Ph O 9d	8% 6a , 21 h	84% ^b
3	Ph ROTO 7e	Ph O FO 9e	8% 6a , 3 d	95% ^b
4	R O 7g	Ph O 9g	10% 6a , 3 d	31%
5	Ph Ph 7f	Ph P	6% 6a , 5 d	no RCM ^c
6	C Ph O R O 7a		16% 6a , 5 d	34%
7	Q Ph N H 7h	Ph O R NH 9h	3% 6a , 18 h	85%
8	Ph Ph N H 7i	Ph Ph Ph 9i	8% 6a , 24 h	63% ^b
9	^Q , Ph N [−] K [−] N [−] 7b	Ph O HN ^{-P} NH 9b	6% 6a , 3 d	36%
10	Q NHCH₂Ph P 7j	Q _R NHCH₂Ph ∮9j	6% 6a , 2 d	43%
11		COOMe 9k	6% 6a , 2 d	80%

Table 1. RCM of dienes **7a**–**k** in the presence of the Ru-catalyst **6a** in CH₂Cl₂ (reflux, 0.02 M)

^aIsolated yields.

^bMixture of diastereomers.

^c100% recovered starting material.

The symmetrical diene **7a** was prepared by the addition of two equivalents of allylic alcohol to phenylphosphonic dichloride in the presence of two equivalents of triethylamine and a catalytic amount of DMAP in 98% yield. Similarly, diene **7b** was prepared in 41% yield using an excess of allylamine. Dienes **7c**, **7d**, **7e** and **7f** were prepared in two steps from phenyldichlorophosphine according to a procedure described in the literature.¹⁵ Addition of two equivalents of the corresponding allylic alcohol in the presence of pyridine followed by a [2,3] sigmatropic Arbusov rearrangement at 130°C of the intermediate phosphonites afforded the expected dienes with non optimised overall yields ranging from 31% to 46%. The major problem of this route is the concomitant formation of the phosphonates 8c-f resulting from the oxidation of the intermediate diallyl phenyl phosphonites. However, the desired

dienes 7c–f could be obtained analytically pure by column chromatography. Dienes 7g, 7h and 7i were prepared from allylphenylphosphinic methyl ester¹⁶ by treatment with PCl₅ followed by addition of the corresponding allylic alcohol or amine in the presence of a catalytic amount of DMAP in dichloromethane. Finally, dienes 7j and 7k were prepared from diallylphosphinic acid ¹⁷ and the corresponding amines via the activated phosphinic acid chloride.

The RCM of dienes 7a-k were carried out in dichloromethane under reflux in the presence of 2-16% (added portionwise, 2 mol%) of the Ru-catalyst 6a and at a concentration of 0.02 M to give the corresponding cyclised products with moderate to excellent yields (Table 1). Under these conditions, RCM of the diene 7c proceeded smoothly to give the phenyl-substituted six-membered oxaphosphinine oxide 9c in excellent yield (entry 1). The cyclic products 9d and 9e possessing a methyl group respectively on position 6 and 3 could also be prepared conveniently and in very high yields (entries 2 and 3). However, the formation of the structural isomer 9g possessing the methyl group on the olefin, was less favourable. Indeed, only 31% of product 9g was isolated after three days in the presence of 10% alkylidene 6a (entry 4). The tetrasubstituted cyclic olefin 9f could not be obtained using this procedure. Indeed, the alkylidene 6a showed no reaction with the disubstituted diene 7f over five days supporting the hypothesis that steric effects are unfavourable for promoting ring closure (entry 5).¹⁸ The cyclisation of the diene 7a including a phosphonate group was surprisingly sluggish and gave only 34% of the seven-membered adduct 9a after five days in the presence of 16% of catalyst. ¹H NMR and mass analysis of the crude mixture showed only unreacted starting material and no trace of dimer (entry 6).

Having established that the formation of differently substituted oxaphosphinine oxides (entries 1 to 4) was feasible, attention was turned towards the compatibility of RCM for the preparation of the corresponding nitrogen containing heterocycles. It was presumed that RCM of dienes including free allylic phosphinamide or phosphonamide NH groups might be problematic as it was previously observed by other groups that cyclisation of subtrates containing free allylic amide NH group could be difficult.¹⁹ Interestingly, the reaction of the diene 7h required only 3% of catalyst 6a for 85% yield of product 9h after 18 h (entry 7). Exposure of the phenyl-substituted diene 7i to the same alkylidene 6a showed lower reactivity in converting this substrate to 9i. Indeed, after 48 h in the presence of 8% of alkylidene 6a, 63% of the expected cyclic product 9i was isolated (entry 8). When the phosphonamide 7b was subjected to RCM condition, only 36% of the desired product 9b could be isolated after three days in the presence of 6% of the Ru-catalyst 6a (entry 9). For this reaction, the NMR and mass spectrum of the crude mixture show only unreacted starting material with no trace of a dimer or other side products. The lower reactivity of substrate 7b supports the observation that the formation of a seven-membered ring combined with the presence of free allylic NH groups is not favourable to the RCM reaction. Finally, we examined the reactivity of phosphinamides 7j and 7k. These dienes with alkylidene 6a afforded the expected 5-membered ring products in 43%

and 80% yield, respectively (entries 10 and 11). The reaction of the tertiary phosphinamide **7k** was more favourable, supporting once more the hypothesis that, in some cases, the catalyst might be inhibited by the complex-forming properties of the phosphinamide NH group.

Conclusion

In summary, the work presented here has established a general strategy based on the RCM reaction for the synthesis of various phosphorus containing five-, six- and seven-membered heterocycles. During the course of this study, it was reported by Hanson et al. that 6-membered allylphosphonamides, five-membered vinylphosphonamides and various five-, six- and seven-membered phosphonates could be prepared by RCM of the corresponding dienes.²⁰ His results are consistent with our observations as he also found that the reactions with substrates including free NH groups might be sluggish and required prolonged reaction time and higher amount of the catalyst. Both papers are complementary and show that the RCM is a valuable synthetic route for a wide range of differently substituted phosphorus containing heterocycles. Application of this methodology to new transition state analogs is in progress in this laboratory.

Experimental

General methods

Allylphenylphosphinic methyl ester,¹⁶ diallylphenylphosphinic acid¹⁷ and ammonium phosphinate²¹ were prepared as described in the literature. The ruthenium alkylidene 6a was purchased from Strem. ¹H NMR and ¹³C NMR spectra were recorded on Bruker DPX-400 and Bruker AM-500 spectrometers. ³¹P NMR spectra were recorded on a Bruker AM-500 spectrometer at 202 MHz and were referenced externally to phosphoric acid ($\delta_P=0$ ppm). Infrared spectra were recorded using a Perkin-Elmer Paragon 1000 FT-IR spectrometer. Mass spectra (m/z) were recorded on a Micromass Platform-I APCI spectrometer. HRMS were performed on a Micromass Autospec 5000 OATof. Thin layer chromatography was performed using Merck aluminium foil backed sheets pre-coated with Kieselgel 60 F₂₅₄. Plates were visualised using UV light or KMnO₄. Column chromatography was performed using Sorbsil[®] C₆₀ H (40–60) silica gel.

Phenylphosphonic acid diallyl ester 7a.²² To a solution of phenylphosphonic dichloride (0.3 ml, 2.12 mmol), a catalytic amount of DMAP and Et₃N (0.65 ml, 4.7 mmol) in CH₂Cl₂ (10 ml) at 0°C was added allylic alcohol (0.32 ml, 4.71 mmol). The mixture was allowed to warm to room temperature and was stirred overnight. The solvent was removed under reduced pressure and the residue was treated with pentane (10 ml). After filtration of the salts, the residue was purified by column chromatography; yield: 0.5 g (98%); $R_{\rm f}$ =0.2 (hexane:EtOAc, 3:2); $\delta_{\rm H}$ (CDCl₃, 400 MHz) 7.86–7.80 (m, 2H), 7.59–7.54 (m, 1H), 7.50–7.45 (m, 2H), 5.98–5.89 (m, 2H), 5.34 (d, 2H, *J*=17.1 Hz), 5.22 (d, 2H, *J*=10.4 Hz), 4.64–4.49 (m, 4H); $\delta_{\rm C}$ (CDCl₃,

101 MHz) 132.8 (d, J=6.8 Hz), 132.6 (d, J=2.8 Hz), 131.8 (d, J=10.1 Hz), 128.5 (d, J=15.0 Hz), 127.8 (d, J=189.5 Hz), 118.0, 66.5 (d, J=5.3 Hz); $\delta_{\rm P}$ (CDCl₃, 202 MHz) 20.9; $\nu_{\rm max}$ (neat, cm⁻¹) 1650, 1249, 1012; m/z (CI⁺, NH₃) 239.0 (M+H⁺).

N,*N*′-**Dially**1-*p*-**pheny**1**phosphonic diamide 7b.** Same procedure as for **7a** with 0.3 ml (2.12 mmol) of phenyl phosphonic dichloride a catalytic amount of DMAP and 3 ml Et₃N (21.2 mmol), 1.6 ml allylamine (21.2 mmol). yield: 0.2 g (41%); $R_{\rm f}$ =0.2 (1% MeOH in EtOAc); $\delta_{\rm H}$ (CDCl₃, 500 MHz) 7.87–7.83 (m, 2H), 7.54–7.51 (m, 1H), 7.48–7.44 (m, 2H), 5.91–5.84 (m, 2H), 5.27 (dd, 2H, *J*=16.8, 1.3 Hz), 5.13 (dd, 2H, *J*=10.5, 1.5 Hz), 3.64–3.58 (m, 4H), 2.56 (br s, 2H); $\delta_{\rm C}$ (CDCl₃, 101 MHz) 136.5 (d, *J*=6.5 Hz), 131.7 (d, *J*=2.5 Hz), 131.5 (d, *J*=9.2 Hz), 124.9 (d, *J*=184.0 Hz), 115.4, 43.2; $\delta_{\rm P}$ (CDCl₃, 202 MHz) 22.0; $\nu_{\rm max}$ (neat, cm⁻¹) 3216, 1644, 1189; *m/z* (CI⁺, NH₃) 237.2 (M+H⁺); HRMS calcd for C₁₂H₁₈ N₂OP (M+H⁺) 237.1157, found 237.1156.

Allylphenylphosphinic acid allyl ester 7c.¹⁵ To a solution of allylic alcohol (1.1 ml, 16.1 mmol) and pyridine (1.3 ml, 16.1 mmol) in Et₂O (30 ml) at 0°C was added a solution of dichlorophenylphosphine (1 ml, 7.4 mmol) in Et₂O (1 ml). The mixture was allowed to warm to room temperature and then refluxed for 1 h. Salts were filtered off and the solvent was removed under reduced pressure. The residue was heated neat to 130°C overnight. The crude mixture was purified by column chromatography; yield: 0.5 g (31%); $R_{\rm f}$ =0.3 (EtOAc); $\delta_{\rm H}$ (CDCl₃, 500 MHz) 7.78–7.76 (m, 2H), 7.57-7.54 (m, 1H), 7.50-7.46 (m, 2H), 5.95-5.88 (m, 1H), 5.81–5.71 (m, 1H), 5.33 (dd, 1H, J=17.1, 2.0 Hz), 5.21 (dd, 1H, J=10.4, 1.5 Hz), 5.15 (dd, 1H, J=10.1, 4.0 Hz), 5.08 (dd, 1H, J=17.1, 5.1 Hz), 4.56 (m, 1H, J=13.5, 5.5 Hz), 4.32 (dd, 1H, J=13.0, 5.5 Hz), 2.86-2.77 (m, 2H); $\delta_{\rm C}$ (CDCl₃, 126 MHz) 133.0 (d, J=7.0 Hz), 132.0, 131.8 (d, J=9.4 Hz), 130.0 (d, J=124.9 Hz), 128.5 (d, J=12.5 Hz), 127.0 (d, J=9.3 Hz), 120.5 (d, J=13.4 Hz), 117.7, 65.0 (d, J=5.5 Hz), 36.0 (d, J=97.0 Hz); δ_P (CDCl₃, 202 MHz) 42.4; ν_{max} (neat, cm⁻¹) 1638, 1233, 1018; m/z (CI^+, NH_3) 223.1 $(M+H^+)$.

But-2-enylphenylphosphinic acid 1-(methyl)allyl ester 7d. Same procedure as for 7c with 1.41 ml (16.3 mmol) of 1-buten-3-ol, 1.31 ml (16.3 mmol) pyridine, 30 ml Et₂O, 1 ml (7.4 mmol) dichlorophenylphosphine; two diastereomers A and B (1:1 ratio); yield: 0.8 g (46%); $R_{\rm f}$ =0.3 (EtOAc: Hexane, 2:1); δ_H (CDCl₃, 400 MHz) (mixture of diastereomers) 7.79-7.71 (m, 2H), 7.53-7.40 (m, 3H), 5.99-5.91 (m, 1H, A or B), 5.79-5.71 (m, 1H, A or B), 5.48–4.98 (m, 4H), 4.98–4.91 (m, 1H, A or B), 4.82–4.77 (m, 1H, A or B), 2.68 (dd, 2H, J=6.4, 17.1 Hz), 1.63–1.59 (m, 3H), 1.44 (d, 3H, J=6.8 Hz, A or B), 1.25 (d, 3H, J=6.0, A or B); $\delta_{\rm C}$ (CDCl₃, 101 MHz) (mixture of diastereomers) 139.0 (d, J=3.9 Hz, A or B), 138.6 (d, J=5.5, A or B), 132.0 (d, J=18.9 Hz), 131.6 (d, J=9.6 Hz), 131.5 (d, J=125.3 Hz), 131.1 (d, J=13.3 Hz), 128.2 (d, J=12.5 Hz), 119.3, 115.4, 73.0 (d, J=6.5 Hz, A or B), 72.5 (d, J=6.3 Hz, A or B), 35.0 (d, J=97.9 Hz), 22.8 (A or B), 22.2 (A or B), 18.0; δ_P (CDCl₃, 202 MHz) 42.1 (A or B), 41.5 (A or B) ν_{max} (neat, cm⁻¹) 1646, 1592, 1227, 1007; *m*/*z* (CI⁺,

NH₃) 251.1 (M+H⁺); HRMS calcd for $C_{14}H_{21}O_2P$ (M+H⁺) 251.1201, found 251.1204.

2-(Methyl)allylphenylphosphinic acid but-2-envl ester (mixture of Z and E isomers) 7e. Same procedure as for 7c with 1.88 ml (22 mmol) of but-2-en-ol, 1.8 ml (22 mmol) of pyridine, 32 ml Et₂O, 1.36 ml (10 mmol) of dichlorophenylphosphine; yield: 1.13 g (45%); $R_f=0.5$ (hexane: EtOAc, 1:1); $\delta_{\rm H}$ (CDCl₃, 400 MHz) (*minor isomer) 7.72-7.66 (m,2H), 7.5-7.46 (m,1H), 7.42-7.38 (m,2H), 5.77-5.64 (m,2H), 5.53 (m,1H), 5.11-4.87 (m,2H), 4.47-4.41 (m,1H), 4.24-4.19 (m,1H), 2.78-2.69 (m,1H), 1.63 (dd, J=6.4, 1.0 Hz, 3H), 1.22 (ddd, J=7.6, 17.2, 1.2 Hz, 3H) (*1.15 (ddd, J=17.6, 1.2 Hz); $\delta_{\rm C}$ (CDCl₃, 101 MHz) (*minor isomer) 134.2 (d, J=6.7 Hz) (*134 (d, J=8.7 Hz)), 132.4 (d, J=8.5 Hz) (*132.3 (d, J=8.5 Hz)), 132.1 (d, J=2.3 Hz), 130.6 (*130.5), 130.4 (d, J=139.5 Hz), 128.1 (d, J=11.7 Hz) (*128.3 (d, J=11.4 Hz)), 126.17 (d, J=7.1 Hz (*126.14 (d, J=6.9 Hz)), 117.4 (d, J=11.9 Hz) $(^*117.6 \text{ (d, } J=11.6 \text{ Hz})), 65.2 \text{ (d, } J=6.5 \text{ Hz}), 39.2 \text{ (d, }$ J=97.7 Hz) (*39.5 (d, J=96.5 Hz)), 17.6, 12.1 (d, J=3.7 Hz) (*12.6 (d, J=3.7 Hz)); δ_P (CDCl₃, 202 MHz) (*minor isomer) 45.6 (*45.5); ν_{max} (neat, cm⁻¹) 1228; m/z (CI^+, NH_3) 251.2 $(M+H^+)$; HRMS calcd for $C_{14}H_{21}O_2P$ (M+H⁺) 251.1201, found 251.1205.

2-(Methyl)allylphenyl-phosphinic acid 2-(methyl)allyl ester 7f. Same procedure as for 7c with 1.36 ml (16.16 mmol) 2-methyl-2-propen-1-ol, of 1.31 ml (16.2 mmol) of pyridine, 30 ml Et₂O, 1 ml (7.4 mmol) of dichlorophenylphosphine; yield: 0.6 g (31%); $R_{\rm f}$ =0.5 (hexane:EtOAc, 2:3); $\delta_{\rm H}$ (CDCl₃, 400 MHz) 7.77–7.72 (m, 2H), 7.53-7.49 (m, 1H), 7.46-7.41 (m, 2H), 4.98 (s, 1H), 4.87 (s, 1H), 4.83-4.81 (m, 1H), 4.64 (d, 1H, J=5.2 Hz), 4.64 (dd, 2H, J=12.4, 5.2 Hz), 4.27 (dd, 2H, J_{AB} =12.4, 6.0 Hz), 2.82–2.68 (m, 2H), 1.77 (t, 3H, J=1.4 Hz), 1.71 (s, 3H); $\delta_{\rm C}$ (CDCl₃, 101 MHz) 141.0 (d, J=7.3 Hz), 136.3 (d, J=9.6 Hz), 132.8 (d, J=1.7 Hz), 132.4 (d, J=9.6 Hz), 130.7 (d, J=124.4 Hz), 129.0 J=12.6 Hz), 116.5 (d, J=11.0 Hz), 113.3, 68.1 (d. (d, J=6.3 Hz), 40.1 (d, J=95.6 Hz), 24.6, 19.7; $\delta_{\rm P}$ (CDCl₃, 202 MHz) 42.1; $\nu_{\rm max}$ (neat, cm⁻¹) 1648, 1230, 1007; m/z (CI^+, NH_3) 251.2 $(M+H^+)$; HRMS calcd for $C_{14}H_{20}O_2P$ (M+H⁺) 251.1201, found 251.1203.

Allylphenylphosphinic acid 2-(methyl)allyl ester 7g. To a solution of 0.5 g of allylphenylphosphinic acid in CH₂Cl₂ (5 ml) was added PCl_5 (0.53 g). The mixture was stirred at room temperature for 24 h. The solvent was removed under reduced pressure and the residue was placed at the vacuum pump for 15 min. To a solution of the crude phosphinic acid chloride (0.3 g) at 0°C in CH₂Cl₂ (10 ml) was then added dropwise a catalytic amount of DMAP, Et₃N (0.42 ml, 3 mmol) and 2-methyl allylic alcohol (0.25 ml, 3 mmol). The mixture was stirred for 72 h. The solvent was removed under reduced pressure and the crude product purified by column chromatography; yield: 80 mg (39%); $R_{\rm f}$ =0.4 (EtOAc); $\delta_{\rm H}$ (CDCl₃, 400 MHz) 7.81–7.76 (m, 2H), 7.597.55 (m,1H), 7.51-7.46 (m, 2H), 5.83-5.71 (m,1H), 5.17-5.06 (m,2H), 5.02 (d, 1H, J=1.2 Hz), 4.92 (d, 1H, J=1.2 Hz), 4.47 (dd, 1H, J=12.6, 6.3 Hz), 4.19 (dd, 1H, J=12.6, 6.6 Hz), 2.82 (dd, J=7.6, 16.8 Hz), 1.75 (s, 3H); $\delta_{\rm C}$ (CDCl₃, 101 MHz) 140.4 (d, J=7.2 Hz), 132.4 (d,

J=2.6 Hz), 131.8 (d, J=9.7 Hz), 129.9 (d, J=125.2 Hz), 128.5 (d, J=12.6 Hz), 127.0 (d, J=13.2 Hz), 112.7, 67.6 (d, J=6.3 Hz), 35.9 (d, J=97.1 Hz), 19.1; $\delta_{\rm P}$ (CDCl₃, 202 MHz) 42.14; $\nu_{\rm max}$ (neat, cm⁻¹) 1648, 1230, 1007; HRMS calcd for C₁₃H₁₈O₂P (M+H⁺) 237.1044, found 237.1045.

Allylphenylphosphinic acid allyl amide 7h. Same procedure as for 7g with 0.5 g of allylphenylphosphinic acid, 5 ml of CH₂Cl₂, 0.53 g of PCl₅ then 1.07 ml (7.6 mmol) of Et₃N, 0.57 ml (7.65 mmol) of allylamine; yield: 232 mg (38%); $R_{\rm f}$ =0.6 (10% MeOH in EtOAc); $\delta_{\rm H}$ (CDCl₃, 500 MHz) 7.87–7.80 (m,2H), 7.56–7.42 (m,3H), 5.93–5.77 (m,2H), 5.24–5.06 (m,4H), 3.57 (m,1H), 3.44 (m, 1H) 2.87 (br s,1H), 2.85–2.70 (m,2H); $\delta_{\rm C}$ (CDCl₃, 126 MHz) 136.2 (d, *J*=8.4 Hz), 132.3 (d, *J*=8.6 Hz), 132.0 (d, *J*=3.4 Hz), 130.9 (d, *J*=125.1 Hz),128.5 (d, *J*=13.1 Hz), 128.3 (d, *J*=9.6 Hz), 120.1 (d, *J*=12.4 Hz), 115.8, 42.7, 36.6 (d, *J*=87.7 Hz); $\delta_{\rm P}$ (CDCl₃, 202 MHz) 31.1; $\nu_{\rm max}$ (neat, cm⁻¹) 3187, 1638, 1176; *m/z* (CI⁺, NH₃) 222.1 (M+H⁺); HRMS calcd for C₁₂H₁₇ONP (M+H⁺) 222.1048, found 222.1041.

Allylphenylphosphinic acid 1-(phenyl)allyl amide 7i. 0.5 g of allylphenylphosphinic acid, 5 ml CH₂Cl₂, 0.53 g of PCl₅ then 1.07 ml (7.6 mmol) of Et₃N, 1.02 g (7.66 mmol) 1-(phenyl)allylamine; yield (mixture of two diastereomers A and B, ratio 1:1): 270 mg (33%); $R_f=0.6$ (10% MeOH in EtOAc); $\delta_{\rm H}$ (CDCl₃, 400MHz) (mixture of two diastereomers A and B, ratio 1:1) 7.86-7.67 (m, 2H), 7.57-7.16 (m, 8H), 6.12-6.05 (m, 1H), 6.00-5.70 (m, 1H), 5.27-5.07 (m, 4H), 4.79-4.74 (m, 1H), 3.33-3.24 (m, 1H), 2.79 (dd, 2H, J=7.6, 17.2, dia A or B), 2.73 (dd, 2H, J=7.6, 17.4, A or B); $\delta_{\rm C}$ (CDCl₃, 101 MHz) (mixture of two diastereomers) 142.0, 141.9, 140.1, 140.1, 139.6, 132.3, 132.2, 131.9, 131.8, 130.6, 128.6, 128.5, 128.4, 128.3, 128.2, 127.4, 127.4, 127.0, 120.2 (d, J=12.3 Hz), 115.6, 56.7, 36.9 (d, J=87.6 Hz, A or B), 36.7 (d, J=88.0 Hz, A or B); δ_P (CDCl₃, 202 MHz) 29.9; ν_{max} (CHCl₃, cm⁻¹) 3019, 1222; *m/z* (CI⁺, NH₃) 298.2 $(M+H^+)$; HRMS calcd for $C_{18}H_{21}NOP$ $(M+H^+)$ 298.1361, found 298.1371.

Diallyl phosphinic acid benzyl amide 7j. To a solution of diallyl phosphinic acid (0.4 g, 2.74 mmol) and a catalytic amount of DMF in CH₂Cl₂ (10 ml) at 0°C was added oxalyl chloride (0.7 ml, 8.02 mmol). After addition, the mixture was allowed to warm to room temperature and was stirred for 1 h. The solvent was removed under reduced pressure and the crude acid chloride was used without purification in the next step. To a solution of the crude phosphinic acid chloride, a catalytic amount of DMAP and Et₃N (1.9 ml, 13.6 mmol) in CH₂Cl₂ (15 ml) at 0°C was added the benzylamine (1.5 ml, 13.7 mmol). The mixture was allowed to warm to room temperature and was stirred for 72 h. The solvent was removed under reduced pressure and the residue was treated with Et₂O (10 ml). After filtration of the salts, the residue was purified by column chromatography; yield: 0.5 g (79%); $R_{\rm f}$ =0.4 (5% MeOH in EtOAc); $\delta_{\rm H}$ (CDCl₃, 400 MHz) 7.35-7.26 (m, 5H), 5.91-5.79 (m, 2H), 5.25-4.17 (m, 4H), 4.21-4.17 (m, 2H), 2.74 (br s, 1H), 2.66 (dd, 4H, J=15.9, 7.5 Hz); δ_{C} (CDCl₃, 101 MHz) 139.7, 128.7, 128.3 (d, J=8.9 Hz), 127.5 (d, J=6.9 Hz), 120.3 (d, J=12.2 Hz), 43.8, 34.4 (d, J=83.8 Hz); $\delta_{\rm P}$ (CDCl₃, 202 MHz) 38.1; $\nu_{\rm max}$ (neat, cm⁻¹) 3182, 1637, 1226; *m/z* (CI⁺, NH₃) 258.1 (M+H⁺).

Diallyl phosphinic acid (S)-2-(methoxycarbonyl)pyrrolidyl amide 7k. Same procedure as for 7j with 0.5 g (3.42 mmol) of diallyl phosphinic acid, a catalytic amount of DMF, 20 ml of CH₂Cl₂, 0.9 ml (10.32 mmol) of oxalyl chloride then a catalytic amount of DMAP, 1.43 ml (10.26 mmol) of Et_3N , 20 ml of CH_2Cl_2 , 1.7 g (10.26 mmol) of the hydrochloride salt of L-proline methyl ester; yield: 0.4 g (41%); R_f=0.4 (10% MeOH in EtOAc); $\delta_{\rm H}$ (CDCl₃, 400 MHz) 5.95–5.80 (m, 2H), 5.25–5.17 (m, 4H), 4.39-4.34 (m, 1H), 3.71 (s, 3H), 3.33-3.29 (m, 2H), 2.74-2.61 (m, 4H), 2.16-2.09 (m, 2H), 2.06-1.85 (m, 2H); $\delta_{\rm C}$ (CDCl₃, 101 MHz) 174.6, 128.4 (d, J=8.6 Hz), 128.0 (d, J=9.1 Hz), 120.0 (d, J=10.9 Hz), 59.0, 52.1, 46.8, 34.7 (d, J=82.4 Hz), 33.3 (d, J=82.7 Hz), 31.1 (d, J=5.5 Hz), 25.2 (d, J=6.1 Hz); δ_{P} (CDCl₃, 202 MHz) 40.7; ν_{max} (neat, cm^{-1}) 1743, 1637, 1210, 920; *m*/*z* (CI⁺, NH₃) 258.1 $(M+H^+)$; HRMS calcd for $C_{12}H_{21}$ NO₃P $(M+H^+)$ 258.1259, found 258.1256.

General procedure for ring-closing metathesis reactions

To a solution of the diene $7\mathbf{a}-\mathbf{k}$ in dry CH₂Cl₂ (0.02 M) was added portionwise (2 mol%) the Grubbs' catalyst. The reaction was refluxed until maximum conversion as shown by TLC or ¹H NMR. The reaction was then concentrated and purified by column chromatography.

2-Phenyl-4,7-dihydro-[1,3,2]dioxaphosphepine 2-oxide 9a. 100 mg (0.4 mmol) of **7a**, 20 ml of CH₂Cl₂, catalyst (42 mg, 12%), 5 d; yield: 31 mg (34%); R_f =0.2 (hexane: EtOAc, 1:1); δ_H (CDCl₃, 400 MHz) 7.90–7.85 (m, 2H), 7.62–7.57 (m, 1H), 7.52–7.47 (m, 2H), 5.81 (t, 2H, J=1.8 Hz), 4.88 (dd, 2H, J_{AB} =15.8 Hz), 4.59 (dd, 2H, J_{AB} =15.5 Hz); δ_C (CDCl₃, 101 MHz) 132.8 (d, J=2.9 Hz), 131.6 (d, J=9.7 Hz), 128.6 (d, J=15.2 Hz) 127.5, 63.6 (d, J=6.7 Hz); δ_P (CDCl₃, 202 MHz) 24.5; ν_{max} (neat, cm⁻¹) 1256; m/z (Cl⁺, NH₃) 211.0 (M+H⁺); HRMS calcd for C₁₀H₁₂O₃P (M+H⁺) 211.0524, found 211.0527.

2-Phenyl-1,3,4,7-tetrahydro-[1,3,2]diazaphosphepine 2-oxide 9b. 77 mg (0.3 mmol) of **7b**, 17 ml of CH₂Cl₂, catalyst (18 mg, 6%), 3 d; yield: 24 mg (36%); $R_{\rm f}$ =0.2 (5% MeOH in EtOAc); $\delta_{\rm H}$ (CDCl₃, 500 MHz) 7.95–7.91 (m, 2H), 7.58–7.55 (m, 1H), 7.51–7.47 (m, 2H), 5.70 (t, 2H, J=2.2 Hz), 3.91 (m, 2H), 3.67 (m, 2H), 3.29 (br, 2H); $\delta_{\rm C}$ (CDCl₃, 126 MHz) 131.8, 131.2 (d, J=9.2 Hz), 128.9, 128.5 (d, J=13.1 Hz), 39.5; $\delta_{\rm P}$ (CDCl₃, 202 MHz) 26.0; $\nu_{\rm max}$ (CHCl₃, cm⁻¹) 3019, 1520, 1210; m/z (CI⁺, NH₃) 209.0 (M+H⁺); HRMS calcd for C₁₀H₁₄ N₂OP (M+H⁺) 209.0844, found 209.0839.

2-Phenyl-3,6-dihydro-[1,2]oxaphosphinine 2-oxide 9c. 100 mg (0.5 mmol) of **7c**, 20 ml of CH₂Cl₂, catalyst (14 mg, 4%), 16 h; yield: 80 mg (92%); $R_{\rm f}$ =0.3 (3% MeOH in EtOAc); $\delta_{\rm H}$ (CDCl₃, 500 MHz) 7.85–7.81 (m, 2H), 7.59–7.56 (m, 1H), 7.51–7.47 (m, 2H), 5.96–5.85 (m, 2H), 5.06–4.99 (m, 1H), 4.76–4.69 (m, 1H), 2.75–2.57 (m, 2H); $\delta_{\rm C}$ (CDCl₃, 126 MHz) 132.6, 131.0 (d, *J*=135.6 Hz), 130.9 (d, *J*=10.6 Hz), 128.6 (d, *J*=13.2 Hz), 126.2 (d, J=16.4 Hz), 120.5 (d, J=9.0 Hz), 66.1 (d, J=7.2 Hz), 25.3 (d, J=89.8 Hz); $\delta_{\rm P}$ (CDCl₃, 202 MHz) 32.1; $\nu_{\rm max}$ (neat, cm⁻¹) 1227, 1068; *m*/*z* (CI⁺, NH₃) 195.0 (M+H⁺); HRMS calcd for C₁₀H₁₂O₂P (M+H⁺) 195.0575, found 195.0575.

2-Phenyl-6-methyl-3,6-dihydro-[1,2]oxaphosphinine **2-oxide 9d.** 120 mg (0.5 mmol) of **7d**, 20 ml of CH₂Cl₂, catalyst (28 mg, 8%), 21 h; two diastereomers A and B (ratio 1:1); yield: 84 mg (84%); $R_{\rm f}$ =0.2 (EtOAc); $\delta_{\rm H}$ (CDCl₃, 400 MHz) 7.83-7.73 (m, 2H), 7.55-7.49 (m, 1H), 7.46-7.41 (m, 2H), 5.89-5.72 (m, 2H), 5.25-5.14 (m, 1H, A or B), 4.88-4.81 (m, 1H, A or B), 2.67-2.41 (m, 2H), 1.51 (d, 3H, J=6.7 Hz, A or B), 1.44 (d, 3H, J=6.8, A or B); $\delta_{\rm C}$ (CDCL₃, 101 MHz) 132.5 (d, J=2.4 Hz, A or B), 132.4 (d, J=2.4 Hz, A or B), 132.0, 131.4 (d, J=13.7 Hz, A or B), 131.2 (d, J=10.3 Hz, A or B), 130.8, 130.5 (d, J=10.1 Hz), 128.6 (d, J=6.8 Hz, A or B), 128.5 (d, J=7.1 Hz, A or B), 120.1 (d, J=9.2 Hz, A or B), 119.5 (d, J=8.3 Hz, A or B), 75.7 (d, J=8.2 Hz, A or B), 71.6 (d, J=7.1 Hz, A or B), 25.2 (d, J=90.5 Hz, A or B), 24.3 (d, J=90.2 Hz, A or B), 22.6 (d, J=3.6 Hz, A or B), 22.1 (d, J=7.2 Hz, A or B); δ_P (CDCl₃, 202 MHz) 32.9 (A or B), 31.1 (A or B); ν_{max} (neat, cm⁻¹) 1226; m/z(CI⁺, NH₃) 209.1 (M+H⁺), 417.1 (2M+H⁺); HRMS calcd for $C_{11}H_{14}O_2P$ (M+H⁺) 209.0731, found 209.0732.

2-Phenyl-3-methyl-3,6-dihydro-[1,2]oxaphosphinine 2-oxide 9e. 100 mg (0.4 mmol) of diene 7e, 20 ml of CH₂Cl₂, catalyst (26 mg, 8%); yield: 79 mg (95%); less polar diastereomer: $R_{\rm f}$ =0.3 (2% MeOH in AcOEt); $\delta_{\rm H}$ (CDCl₃, 250 MHz) 7.91–7.82 (m, 2H), 7.63–7.46 (m, 3H), 5.86–5.64 (m, 2H), 5.05–4.94 (m, 1H), 4.75 (m, 1H), 2.75 (m, 1H), 1.30 (dd, J=7.45, 16.8 Hz); $\delta_{\rm C}$ (CDCL₃, 62.9 MHz) 133.0, 131.9 (d, J=9.9 Hz), 130.4 (d, J=133.7 Hz), 128.97 (d, J=12.8 Hz), 127.96 (d, J=8.1 Hz), 125.3 (d, J=15.2 Hz), 66.1 (d, J=6.9 Hz), 29.4 (d, J=91.5 Hz), 14.4; $\delta_{\rm P}$ (CDCl₃, 202 MHz) 36.2 $\nu_{\rm max}$ (neat, cm⁻¹) 1226; more polar diastereomer: $R_f=0.23$ (2%) MeOH in AcOEt); $\delta_{\rm H}$ (CDCl₃, 250 MHz) 7.86–7.76 (m, 2H), 7.59-7.42 (m, 3H), 5.94-5.77 (m, 2H), 5.08-4.95 (m, 1H), 4.85-4.70 (m, 1H), 2.65 (m, 1H), 1.00 (dd, J=7.48, 18.9 Hz); δ_{C} (CDCL₃, 62.9 MHz) 132.8, 132.3 (d, J=9.4 Hz), 129.4 (d, J=133.0 Hz), 129.0 (d, J=13.5 Hz), 128.7 (d, J=6.2 Hz), 125.5 (d, J=15.0 Hz), 66.6 (d, J=7.3 Hz), 30.7 (d, J=89.4 Hz), 16.7; δ_P (CDCl₃, 202 MHz) 38.3; ν_{max} (neat, cm⁻¹) 1226; HRMS calcd for $C_{11}H_{14}O_2P(M+H^+)$ 209.0730, found 209.0731.

2-Phenyl-4,5-dimethyl-3,6-dihydro-[1,2]oxaphosphinine 2-oxide 9f. 50 mg (0.2 mmol) of diene **7f**, 10 ml of CH_2Cl_2 , catalyst (9 mg, 6%), 5 d; yield: 0% (100% recovered diene **7f**).

2-Phenyl-5-methyl-3,6-dihydro-[1,2]oxaphosphinine 2-oxide 9g. 75 mg (0.32 mmol) of diene **7g**, 15 ml of CH₂Cl₂, catalyst (25 mg, 10%), 3 d; yield: 21 mg (31%); $R_{\rm f}$ =0.16 (AcOEt); $\delta_{\rm H}$ (CDCl₃, 400 MHz) 7.86–7.80 (m, 2H), 7.59–7.55 (m, 1H), 7.51–7.47 (m, 2H), 5.60 (dxm, J=23.2 Hz, 2H), 4.90–4.85 (m, 1H), 4.58–4.50 (m, 1H), 2.66–2.56 (m, 2H), 1.72 (s, 3H); $\delta_{\rm C}$ (CDCl₃, 100.6 MHz) 133.0 (d, J=15.3 Hz), 132.5, 130.9 (d, J=11.4 Hz), 130.8 (d, J=135.7 Hz), 128.6 (d, J=13.0 Hz), 114.9 (d, J=9.2 Hz), 68.7 (d, J=11.5 Hz), 25.0 (d, J=86.1 Hz), 19.3; $\delta_{\rm P}$ (CDCl₃, 202 MHz) 32.3; $\nu_{\rm max}$ (neat, cm⁻¹) 1227, 1068; m/z (Cl⁺, NH₃) 195.0 (M+H⁺); HRMS calcd for C₁₁H₁₄O₂P (M+H⁺) 209.0730, found 209.0731.

2-Phenyl-3,6-dihydro-1H-[1,2]azaphosphinine 2-oxide 9h. 65 mg (0.3 mmol) of diene **7h**, 15 ml of CH₂Cl₂, catalyst (7.5 mg, 3%), 18 h; yield: 46.6 mg (85%); $R_{\rm f}$ =0.3 (10% MeOH in EtOAc); $\delta_{\rm H}$ (CDCl₃, 500 MHz) 7.86–7.80 (m, 2H), 7.55–7.51 (m, 1H), 7.48–7.43 (m, 2H), 5.86– 5.76 (m, 2H), 4.15–3.82 (m, 1H), 3.89–3.82 (m, 1H), 2.99 (br, 1H), 2.72–2.52 (m, 2H); $\delta_{\rm C}$ (CDCl₃, 126 MHz) 133.7 (d, *J*=123.6 Hz), 131.9 (d, *J*=3.3 Hz), 131.2 (d, *J*=9.4 Hz), 128.5 (d, *J*=12.4 Hz), 126.3 (d, *J*=16.0 Hz), 119.9 (d, *J*=9.4 Hz), 43.2 (d, *J*=3.8 Hz), 26.8 (d, *J*=88.9 Hz); $\delta_{\rm P}$ (CDCl₃, 202 MHz) 21.5; $\nu_{\rm max}$ (CHCl₃, cm⁻¹) 3019, 1520, 1220; *m/z* (CI⁺, NH₃) 194.1 (M+H⁺); HRMS calcd for C₁₀H₁₃NOP (M+H⁺) 194.0735, found 194.0736.

2,6-Diphenyl-3,6-dihydro-1H-[1,2]-azaphosphinine 2-oxide 9i. 100 mg (0.3 mmol) of **7i**, 20 ml of CH₂Cl₂, catalyst (22 mg, 8%), 24 h; yield: 57 mg (63%); $R_{\rm f}$ =0.4 (EtOAc); $\delta_{\rm H}$ (CDCl₃, 400 MHz) 7.91–7.86 (m, 2H), 7.56–7.47 (m, 4H), 7.39–7.35 (m, 2H), 7.31–7.27 (m, 2H), 5.06–5.05 (m, 1H), 3.15 (s, 1H), 2.76–2.72 (m, 2H), 5.87–5.77 (m, 2H); $\delta_{\rm c}$ (CDCl₃, 101 MHz) 142.5, 131.9, 130.9, 130.7, 128.9, 128.5 (d, *J*=12.5 Hz), 128.0, 127.0, 118.7 (d, *J*=8.2 Hz), 59.8 (d, *J*=3.1 Hz), 26.4 (d, *J*=88.6 Hz); $\delta_{\rm P}$ (CDCl₃, 202 MHz) 21.2; $\nu_{\rm max}$ (CHCl₃, cm⁻¹) 3020, 1522, 1210; *m*/*z* (CI⁺, NH₃) 270.1 (M+H⁺), 539.1 (2M+H⁺); HRMS calcd for C₁₆H₁₇NOP (M+H⁺) 270.1048, found 270.1042.

1-Benzylamino-3-phospholene 1-oxide 9j.²³ 100 mg (0.43 mmol) of **7j**, 20 ml of CH₂Cl₂, catalyst (21 mg, 6%), 2 d; two rotamers A and B (ratio 2:1); yield: 38 mg (43%); $R_{\rm f}$ =0.1 (5% MeOH in EtOAc); $\delta_{\rm H}$ (CDCl₃, 400 MHz) 7.36–7.27 (m, 5H), 5.92 (d, 2H, *J*=31.7 Hz, A), 5.78 (d, 2H, *J*=30.4 Hz, B), 4.18 (d, 2H, *J*=9.2 Hz), 3.92 (s, 2H, B), 2.45 (dAB, 4H, *J*=12.0 Hz, J_{AB} =17.2 and 17.9 Hz, A), 2.02 (d, 4H, *J*=12.7 Hz, B); $\delta_{\rm C}$ (CDCl₃, 126 MHz) 134.1, 128.8 (d, *J*=8.9 Hz), 128.6 (d, *J*=21.9 Hz), 127.6, 127.5 (d, *J*=14.9 Hz), 44.4 (s, A or B), 43.0 (s, A or B), 30.8 (d, *J*=84.3 Hz), 29.7; $\delta_{\rm P}$ (CDCl₃, 202 MHz) 64.2, 63.7; $\nu_{\rm max}$ (CHCl₃, cm⁻¹) 3019, 1217; *m/z* (CI⁺, NH₃) 208.1 (M+H⁺).

1-[(S)-2-(methoxycarbonyl)pyrrolidino]-3-phospholene 1-oxide 9k. 100 mg (0.4 mmol) of diene **7k**, 20 ml of CH₂Cl₂, catalyst (21 mg, 6%), 4 d; yield: 71 mg (80%); R_f=0.3 (10% MeOH in EtOAc); $\delta_{\rm H}$ (CDCl₃, 400 MHz) 6.00–5.84 (m, 2H), 4.37 (dt, 1H, *J*=3.2, 8.4 Hz), 3.70 (s, 3H), 3.13 (td, 2H, *J*=6.8, 3.2 Hz), 2.68–2.60 (m, 1H), 2.44 (d, 3H, *J*=12.4 Hz), 2.21–2.02 (m, 2H), 1.97–1.89 (m, 2H); $\delta_{\rm C}$ (CDCl₃, 101 MHz) 174.6, 127.7 (d, *J*=14.7 Hz), 126.7 (d, *J*=15.0 Hz), 59.4 (d, *J*=2.3 Hz), 52.1, 46.4 (d, *J*=3.9 Hz), 30.9 (d, *J*=6.4 Hz), 29.7 (d, *J*=82.1 Hz), 29.6 (d, *J*=83.6 Hz), 25.1 (d, *J*=6.8 Hz); $\delta_{\rm P}$ (CDCl₃, 202 MHz) 65.6; $\nu_{\rm max}$ (neat, cm⁻¹) 1741, 1613; *m/z* (CI⁺, NH₃) 230.1 (M+H⁺); HRMS calcd for C₁₀H₁₆NO₃P (M+H⁺) 230.0946, found 230.0944.

Acknowledgements

We would like to thank Dr R. Procter for recording the HRMS spectra, E. McGuinness for recording the NMR spectra and Dr S. Ramcharitar for helpful suggestions on this manuscript.

References

1. Brassfield, H. A.; Jacobson, R. A.; Verkade, J. G. *J. Am. Chem. Soc.* **1975**, 4143; Gilard, V.; Martino, R.; Malet-Martino, M.; Niemeyer, U.; Pohl, J. *J. Med. Chem.* **1999**, *42*, 2542.

2. Darrow, J. W.; Drueckhammer, D. G. J. Org. Chem. **1994**. 59, 2976–2985; Hannessian, S.; Galeotti, N.; Rosen, P.; Olva, G.; Babu, S. *Bioorg. Med. Chem.* **1994**, 23, 2763.

3. Rademacher, T. W.; Parekh, R. B.; Dwek, R. A. Ann. Rev. Biochem. **1988**, *57*, 785.

4. Morita, I.; Kunimoto, K.; Tsuda, M.; Tada, S.-I.; Kise, M.; Kimura, K. Chem. Pharm. Bull. 1987, 35, 4144.

5. Hewitt, D. G.; Newland, G. L. Aust. J. Chem. 1977, 30, 579.

6. Bartlett, P. A.; Hanson, J. E.; Gannousis, P. P. J. Org. Chem.

1990, *55*, 6268; Morgan, B. P.; Scholtz, J. M.; Ballinger, M. D.; Lipkin, I. D.; Bartlett, P. A. J. Am. Chem. Soc. **1991**, *113*, 297;

Mader, M. M.; Bartlett, P. A. Chem. Rev. 1997, 97, 1281.

7. Stewart, J. D.; Liotta, L. J.; Benkovic, S. J. Acc. Chem. Res. **1993**, 26, 396; Benkovic, S. J.; Napper, A. D.; Lerner, R. A. Proc. Nat. Acad. Sci. U.S.A. **1987**, 85, 5355.

8. For reviews on RCM: (a) Blechert, S.; Schuster, M. Angew. Chem. Int. Ed. Engl. **1997**, *36*, 2037. (b) Schmalz, H. G. Angew. Chem. Int. Ed. Engl. **1995**, *34*, 1833. (c) Grubbs, R. H.; Chang, S. Tetrahedron **1998**, *54*, 4413; Armstrong, S. K. J. Chem. Soc. Perkin Trans. I **1998**, 371. (a) Nguyen, S. T.; Grubbs, R. H.; Ziller, J. W. J. Am. Chem. Soc.
1993, 115, 9858. (b) Nguyen, S. T.; Johnson, L. K.; Grubbs, R. H.;
Ziller, J. W. J. Am. Chem. Soc. 1992, 114, 3974. (c) Schwab, P.;
Grubbs, R. H.; Ziller, J. W. J. Am. Chem. Soc. 1996, 118, 100.

(a) Bazan, G. C.; Oskam, J. H.; Cho, H.-N.; Park, L. Y.;
Schrock, R. R. J. Am. Chem. Soc. **1991**, 113, 6899. (b) Schrock,
R. R.; Murdzek, J. S.; Bazan, G. C.; Robbins, J.; DiMare, M.;
O'Regan, M. J. Am. Chem. Soc. **1990**, 112, 3875.

11. Leconte, M.; Jourdan, I.; Pagano, S.; Lefebvre, F.; Basset, J.-M. J. Chem. Soc., Chem. Commun. **1995**, 857.

12. Hanson, P. R.; Stoinova, D. S. Tetrahedron Lett. 1998, 39, 3939.

13. Bujard, M.; Gouverneur, V.; Mioskowski, C. J. Org. Chem. 1999, 64, 2119.

14. Trevitt, M.; Gouverneur, V. Tetrahedron Lett. 1999, 40, 7333.

15. Pudovik, A. I.; Aladzheva, I. M.; Spirina, I. V. J. Gen. Chem. USSR (English Trans.) **1967**, *37*, 656.

16. Preparation of allyl phenyl phosphinic acid methyl ester: Patent; Shell Devel. Co., US 2711403; B.

17. Preparation of diallylphenylphosphinic acid: Majewski, P. *Phosphorus, Sulfur, Silicon* **1989**, *45*, 151; Boyd, E. A.; Regan, A. C. *Tetrahedron Lett.* **1994**, *35*, 4223.

18. Kirkland, T. A.; Grubbs, R. H. J. Org. Chem. 1997, 62, 7310.

19. Fu, G. C.; Nguyen, S. T.; Grubbs, R. H. J. Am. Chem. Soc. **1993**, *114*, 9856; Huwe, C. M.; Kiehl, O. C.; Blechert, S. Synlett **1996**, 65.

20. Hanson, P. R.; Stoinova, D. S. Tetrahedron Lett. 1999, 40, 3297.

21. Boyd, E. A.; James, K.; Regan, A. C. *Tetrahedron Lett.* **1992**, *33*, 813.

- 22. Nitta, Y.; Arakawa, Y. Chem. Pharm. Bull. 1986, 34, 3121.
- 23. Kenn, Z., et al., Anorg. Allg. Chem. 1979, 452, 176.